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Bibliography of Electron and Photon Cross Sections with  
Atoms and Molecules

Published in the 20<sup>th</sup> Century

– Argon –

M. Hayashi

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# Bibliography of Electron and Photon Cross Sections

with Atoms and Molecules

Published in the 20<sup>th</sup> Century

—— Argon ——\*

Makoto Hayashi

(Gaseous Electronics Institute)

A bibliography of original and review reports of experiments or theories of electron and photon cross sections and also electron swarm data are presented for atomic or molecular species with specified targets. These works covered 17 atoms and 51 molecules. The present bibliography is only for argon (Ar). About 1,960 papers were compiled. A comprehensive author index is included. The bibliography covers the period 1921 through 2000 for Ar. Finally, author's recommended Ar electron collision cross section set is given in numerical tables.

Keywords : Ar atom, collision cross section, electron, elastic scattering, electronic excitation, ionization, photon, photoabsorption, photodissociation, photoexcitation, photoionization, electron swarm, drift velocity, diffusion coefficient, ionization coefficient, excitation and ionization energies, transition probability, lifetimes of excited states

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# Introduction

## History

This bibliography is the result of a continuing literature survey which was begun around 1970 and originally encompassed only electron collision cross section and electron swarm data. The organization responsible for undertaking this survey is Nagoya Institute of Technology, Nagoya. From 1994, the work continued to Gaseous Electronics Institute, Nagoya. In 1997, the collection of photon cross section references was begun. The search for references in both cases was retrospective and included all papers reporting measurements, calculations or reviews of such cross sections and electron swarm data.

## Scope

This bibliography contains references to original research papers which report experiments or theoretical calculations of cross sections for electron and photon collisions with argon (Ar) atom. The review papers on this subject are also included.

Some argon cluster papers are included. Some conference reports, company or agency reports and Ph.D. thesis are included. Argon ion papers and positron collision papers are not included in principle.

Papers reporting the following data are included.

For electron collision cross sections :

- 1) elastic scattering
- 2) electronic excitation
- 3) ionization
- 4) grand total scattering (sum of elastic and inelastic cross sections)
- 5) metastable argon
- 6) electron swarm parameters (drift velocity, diffusion coefficient)
- 7) excitation and ionization coefficients

For photon collision cross section :

- 1) photoabsorption
- 2) photoexcitation and fluorescence
- 3) photodissociation
- 4) photoionization

For some related data :

- 1) excitation and ionization energies
- 2) transition probabilities
- 3) lifetimes of excited states
- 4) others

The energy range for electron cross section data is usually 0 -10 keV, but some higher electron energy papers are included. The wavelength range for photon cross section data is from microwaves to X-rays. Most papers are concerned with infrared, visible and ultraviolet ray region.

The bibliography includes the papers published in the 20<sup>th</sup> century, from 1901 to 2000. But the oldest paper in this list is given by C. Ramsauer (1921). So for this argon bibliography, published papers from 1921 to 1999 are compiled in alphabetical order of the first author's surname of the paper. References published in 2000 and plus some old papers found very recently after compilation are added as "Addenda of References for Ar". In total, about 1,960 papers are compiled in the argon bibliography.

## Organization

This report consists of four parts : introduction, the bibliography and its addenda, author index and an electron collision cross section set recommended by the author.

## Bibliography

In this section the complete citation for all references are given. At first following classifications are shown :

- E : Elastic collision
- EX : electronic EXcitation
- I : Ionization
- ME: MEtastable argon
- S : electron Swarm
- O : Others (photon cross sections and others)

All author's initials and surname, journal name, volume, inclusive pages and year of publication are given as well as the title, and some additional information in the square bracket [ ]. E and T in the square bracket mean experiment and theory.

## Author Index

In this section all authors are listed alphabetically by surname. After each author's name is a list of page numbers indicating which references he or she authored or coauthored.

### Electron Collision Cross Section Set for Argon (Ar)

Electron collision cross section set of elastic, electronic excitation and ionization collision for argon recommended by the author are given by the tables. Final conclusions are given and detailed discussions are not shown here. Elastic total and elastic momentum transfer cross sections are obtained from author's recommended elastic differential cross sections. These values of the set are not final, in other words, tentative values.

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Institute for Molecular Science, Okazaki

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References for Ar (1900 - 1999)

(Argon)

[Inert gas. Noble gas]

E : Elastic collision,           EX : Electronic excitation,  
I : Ionization,                QT : Grand total cross section,  
ME : Metastable argon,        S : Swarm,     O : The others.  
 $\alpha$  : Ionization coefficient [ ] : Additional informations,  
                                  E : Exp.,   T : Theory.

The oldest paper in this list is given by C. Ramsauer (1921).

- QT W. H. Aberth : Thesis, New York University, 1-57 (1963)  
Measurement of total cross sections for the scattering of low energy  
electrons by argon and molecular oxygen and nitrogen. [E, Ar, O<sub>2</sub>, N<sub>2</sub>]
- QT W. Aberth, G. Sunshine and B. Bederson : in Atomic Collision Processes, North-  
Holland, 3rd ICPEAC, London 53-58 (1964)  
Absolute measurement of total cross sections for the scattering of low  
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Ionization and breakdown in argon. [E, Ar; 15 - 3600 Td]
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Electron Collision Cross Section Set for Argon (Ar)

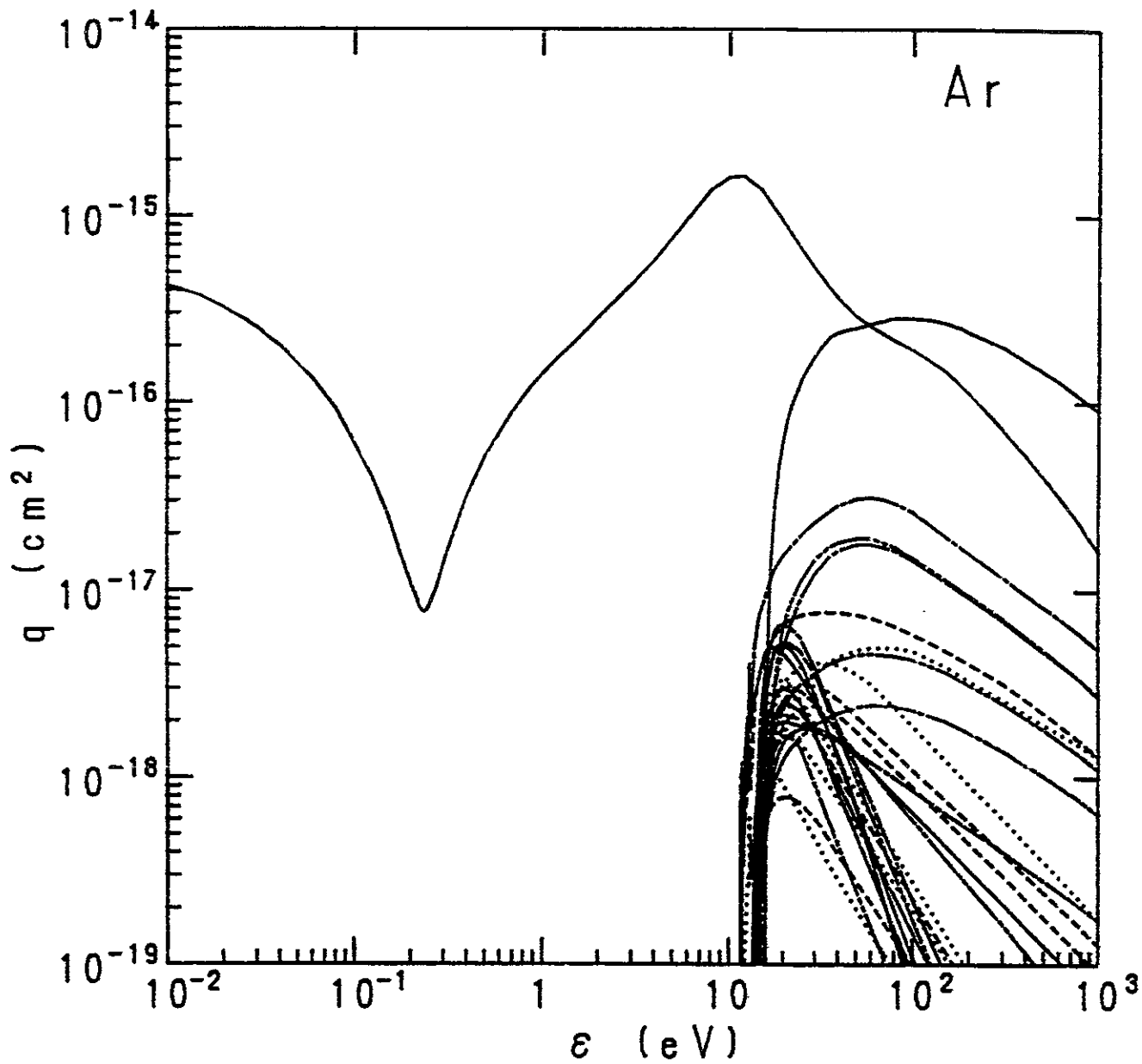


Figure 1. Electron collision cross section set for Ar.



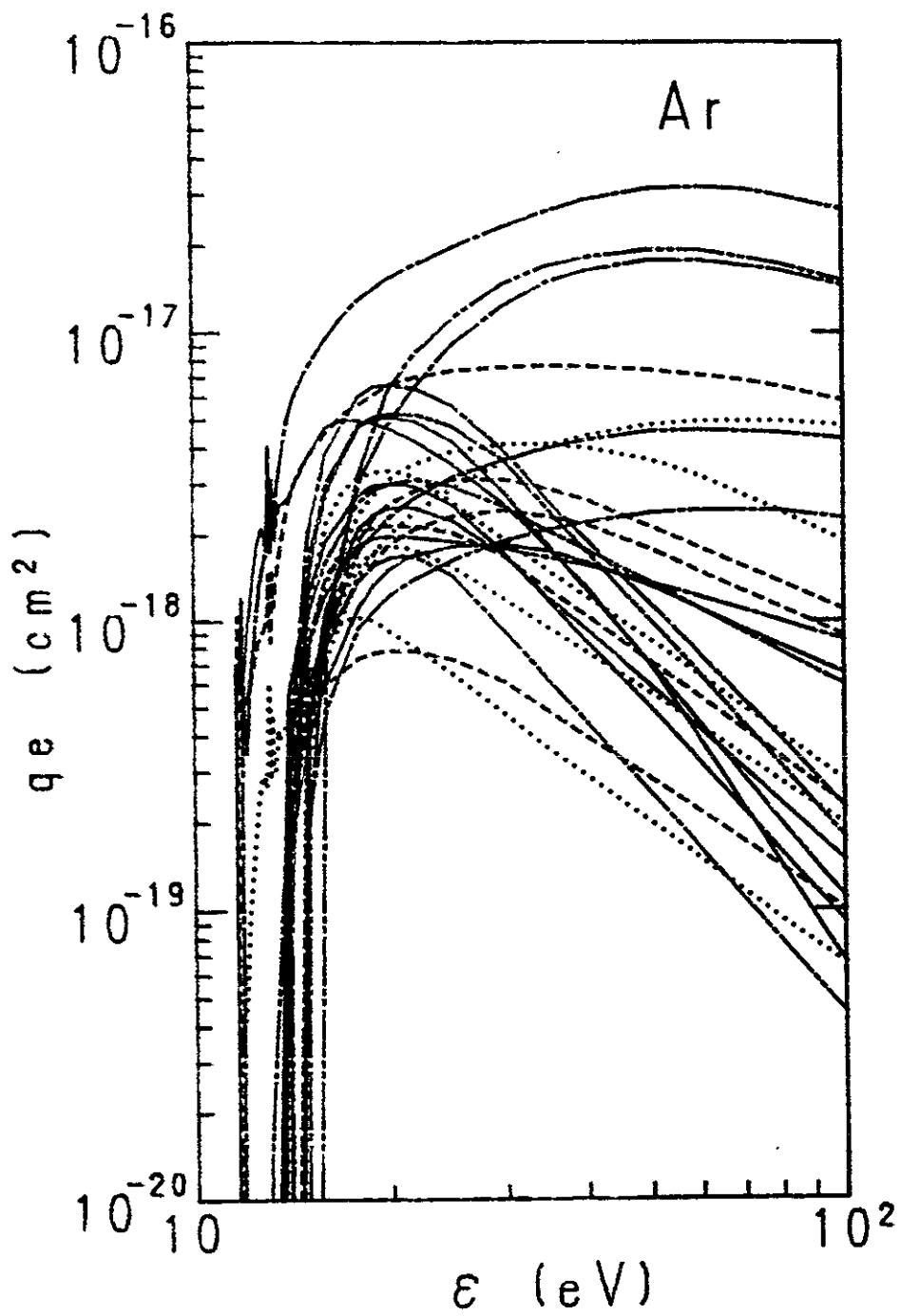


Figure 2. Electronic excitation cross sections for Ar.

Table 1. Energy levels of electronic excitation for Ar.

	Level No.			energy (eV)	Paschen notation	Designation
	This work	Chutjian	Eggarter			
					1 P <sub>0</sub>	3p <sup>6</sup>
1	1	1	SM*	11.55	1 S <sub>5</sub>	4s[3/2] <sub>2</sub>
2	2	2	S 1**	11.62	1 S <sub>4</sub>	4s[3/2] <sub>1</sub>
3	3	3	SM*	11.72	1 S <sub>3</sub>	4s'[1/2] <sub>0</sub>
4	4	4	S 1**	11.83	1 S <sub>2</sub>	4s'[1/2] <sub>1</sub>
5	5	5	S 4 p	12.91	2 P <sub>10</sub>	4p[1/2] <sub>1</sub>
6	6	6		13.08	2 P <sub>9</sub>	4p[5/2] <sub>3</sub>
7	7	7		13.09	2 P <sub>8</sub>	4p[5/2] <sub>2</sub>
8	8	8		13.15	2 P <sub>7</sub>	4p[3/2] <sub>1</sub>
9	9	9		13.17	2 P <sub>6</sub>	4p[3/2] <sub>2</sub>
10	10	10		13.27	2 P <sub>5</sub>	4p[1/2] <sub>0</sub>
11	11	11		13.28	2 P <sub>4</sub>	4p'[3/2] <sub>1</sub>
12				11	13.03	2 P <sub>3</sub>
13	12	12		13.33	2 P <sub>2</sub>	4p'[1/2] <sub>1</sub>
14	13	13		13.47	2 P <sub>1</sub>	4p'[1/2] <sub>0</sub>
15	14	14	13.84	3 d <sub>6</sub>	3d[1/2] <sub>0</sub>	
16	15	15	13.86	3 d <sub>5</sub>	3d[1/2] <sub>1</sub>	
17			15	13.90	3 d <sub>3</sub>	3d[3/2] <sub>2</sub>
18	16	16	13.98	3 d <sub>4</sub> '	3d[7/2] <sub>4</sub>	
19	17	17	14.01	3 d <sub>4</sub>	3d[7/2] <sub>3</sub>	
20	18	18	14.06	3 d <sub>1</sub> ''	3d[5/2] <sub>2</sub>	
21	19	19	14.07	2 S <sub>5</sub>	5s[3/2] <sub>2</sub>	
22			19	14.10	3 d <sub>1</sub> '	3d[5/2] <sub>3</sub>
23	20	20	14.09	2 S <sub>4</sub>	5s[3/2] <sub>1</sub>	
24			20	14.15	3 d <sub>2</sub>	3d[3/2] <sub>1</sub>
25	21	21	14.21	3 S <sub>1</sub> ''''	3d'[5/2] <sub>2</sub>	
26	22	22	14.24	3 S <sub>1</sub> '''	3d'[3/2] <sub>2</sub>	
27			22	14.23	2 S <sub>1</sub> ''	3d'[5/2] <sub>3</sub>
28	23	23	14.24	2 S <sub>3</sub>	5s'[1/2] <sub>0</sub>	
29			23	14.25	2 S <sub>2</sub>	5s'[1/2] <sub>1</sub>
30	23	23	14.30	3 S <sub>1</sub> '	3d'[3/2] <sub>1</sub>	
	24		S 3	14.71		
	25		S 4	15.20		

\* SM : Sum of metastable states, 1 and 3.

\*\* S1 : Sum of 2 and 4 states.

Table 2. Electron collision cross section set for argon, recommended by the author in 1992.  $q_m$  is the elastic momentum transfer,  $q_e$  of 1 to 25 are the electronic excitation, and  $q_i$  is the ionization cross sections. I would like to improve these cross section values slightly.

$\epsilon$ (eV)	$q_m$ ( $10^{-16} \text{cm}^2$ )				
		200	1.12	13.24	0.0209
		250	0.873	13.28	0.0197
		300	0.713	13.3	0.0194
0	6.30	400	0.518	13.33	0.0191
0.01	4.20	500	0.397	13.36	0.0197
0.012	4.00	600	0.315	13.38	0.0206
0.015	3.69	800	0.219	13.40	0.0213
0.02	3.20	1000	0.161	13.44	0.0220
0.025	2.80	10000	0.00	13.48	0.0261
0.03	2.48			13.5	0.0245
0.035	2.20			13.52	0.0237
0.04	1.98	$\epsilon$	$q_{e1}$		
0.05	1.59	(eV)	( $10^{-16} \text{cm}^2$ )	13.6	0.0251
0.06	1.32			14.0	0.0265
0.08	0.913	11.55	0.00	15.0	0.0369
0.10	0.599	11.60	0.00543	16	0.0466
0.125	0.391	11.61	0.00643	17	0.0500
0.15	0.255	11.62	0.00693	18	0.0495
0.175	0.162	11.63	0.00704	19	0.0482
0.20	0.108	11.64	0.00699	20	0.0464
0.225	0.0808	11.65	0.00623	22	0.0418
0.23	0.0781	11.66	0.00633	25	0.0347
0.235	0.0763	11.67	0.00663	30	0.0253
0.24	0.0765	11.68	0.00665	35	0.0180
0.245	0.0786	11.69	0.00633	40	0.0132
0.25	0.0816	11.70	0.00603	50	0.00694
0.275	0.103	11.71	0.00573	60	0.00375
0.3	0.139	11.72	0.00577	70	0.00222
0.35	0.216	11.73	0.00586	80	0.00141
0.4	0.310	11.75	0.00714	100	0.000652
0.5	0.507	11.77	0.00815	150	0.000170
0.6	0.690	11.80	0.00832	175	0.000100
0.7	0.888	11.81	0.00840	200	0.00
0.8	1.07	11.84	0.0120		
0.9	1.25	11.88	0.00681	$\epsilon$	$q_{e2}$
1.0	1.42	11.90	0.00626	(eV)	( $10^{-16} \text{cm}^2$ )
1.2	1.70	11.94	0.00644		
1.5	2.09	12.0	0.00704	11.62	0.00
2.0	2.84	12.1	0.00889	11.68	0.0104
2.5	3.56	12.2	0.0105	11.70	0.00736
3.0	4.31	12.4	0.0150	11.74	0.00568
4.0	5.78	12.6	0.0191	11.76	0.00541
5.0	7.68	12.7	0.0204	11.80	0.00499
6.0	9.71	12.74	0.0208	11.81	0.00494
8.0	13.7	12.76	0.0210	11.85	0.00490
10	16.1	12.8	0.0208	11.90	0.00495
12	16.4	12.86	0.0196	11.94	0.00500
15	13.7	12.90	0.0184	12.00	0.00520
20	9.30	12.91	0.0183	12.04	0.00530
25	6.68	12.94	0.0186	12.1	0.00550
30	5.16	13.0	0.0222	12.2	0.00570
40	3.59	13.05	0.0405	12.4	0.00730
50	2.90	13.08	0.0267	12.5	0.00790
60	2.54	13.10	0.0213	12.6	0.00850
80	2.14	13.13	0.0184	12.65	0.00860
100	1.91	13.16	0.0214	12.7	0.00880
120	1.71	13.19	0.0262	12.74	0.00900
150	1.46	13.21	0.0308	12.8	0.00920
		13.23	0.0221	12.9	0.00970

Ar

13.00	0.0103	12.60	0.00248	11.90	0.00630
13.05	0.0147	12.70	0.00276	11.94	0.00493
13.08	0.0101	12.74	0.00287	12.00	0.00357
13.10	0.00940	12.77	0.00291	12.04	0.00340
13.13	0.00840	12.80	0.00292	12.1	0.00400
13.16	0.0103	12.86	0.00282	12.2	0.00570
13.20	0.0148	12.90	0.00268	12.4	0.00870
13.23	0.0139	12.91	0.00265	12.5	0.0106
13.25	0.0122	12.94	0.00272	12.6	0.0125
13.29	0.0112	13.0	0.00332	12.65	0.0133
13.30	0.0115	13.05	0.00610	12.7	0.0140
13.4	0.0138	13.08	0.00407	12.74	0.0145
13.5	0.0158	13.10	0.00327	12.8	0.0154
13.82	0.0219	13.13	0.00282	12.9	0.0167
14.0	0.0255	13.16	0.00331	13.0	0.0184
14.25	0.0292	13.19	0.00406	13.05	0.0266
14.5	0.0322	13.21	0.00477	13.08	0.0186
15.0	0.0373	13.23	0.00344	13.1	0.0174
16	0.0449	13.24	0.00325	13.13	0.0175
17	0.0520	13.28	0.00310	13.16	0.0192
18	0.0565	13.3	0.00305	13.2	0.0278
19	0.0608	13.33	0.00301	13.23	0.0261
20	0.0658	13.36	0.00312	13.25	0.0229
22	0.0700	13.38	0.00330	13.29	0.0213
25	0.0740	13.40	0.00341	13.3	0.0218
30	0.0760	13.44	0.00354	13.4	0.0265
35	0.0770	13.48	0.00421	13.5	0.0306
40	0.0765	13.50	0.00399	13.82	0.0432
50	0.0743	13.52	0.00385	14.0	0.0505
60	0.0715	13.6	0.00396	14.25	0.0580
70	0.0685	14.0	0.00446	14.5	0.0648
80	0.0650	15.0	0.00641	15.0	0.0757
100	0.0585	16	0.00837	16	0.0947
150	0.0480	17	0.0100	17	0.114
200	0.0410	18	0.0103	18	0.130
300	0.0320	19	0.0100	19	0.142
400	0.0262	20	0.00923	20	0.154
500	0.0222	22	0.00800	22	0.172
800	0.0156	25	0.00650	25	0.197
1000	0.0131	30	0.00475	30	0.233
10000	0.00	35	0.00360	35	0.263
		40	0.00290	40	0.288
		50	0.00200	50	0.313
		60	0.00149	60	0.318
		70	0.00116	70	0.311
		80	0.000930	80	0.296
		100	0.000656	100	0.266
		150	0.000339	150	0.197
		200	0.000212	200	0.160
		300	0.000110	300	0.119
		317	0.000100	350	0.106
		400	0.00	400	0.0963
				500	0.0822
				800	0.0580
				1000	0.0490
				10000	0.00
$\epsilon$	$qe3$	$\epsilon$	$qe4$	$\epsilon$	$qe5$
(eV)	( $10^{-16}cm^2$ )	(eV)	( $10^{-16}cm^2$ )		
11.72	0.00	11.83	0.00		
11.75	0.000202	11.85	0.00770		
11.77	0.000420				
11.80	0.000378				
11.81	0.000400				
11.84	0.000660				
11.88	0.000431				
11.90	0.000418				
11.94	0.000473				
12.00	0.000579				
12.10	0.000837				
12.20	0.00108				
12.40	0.00174				

Argon 2

Ar

(eV)	( $10^{-16}\text{cm}^2$ )	22	0.0211	40	0.0157
		25	0.0232	50	0.0137
12.91	0.00	30	0.0243	60	0.0122
13	0.000140	35	0.0233	80	0.0100
14	0.00300	40	0.0218	100	0.00860
15	0.00590	50	0.0179	150	0.00650
16	0.00870	60	0.0149	200	0.00535
18	0.0140	80	0.0112	300	0.00405
19	0.0156	100	0.00892	500	0.00280
20	0.0166	150	0.00600	800	0.00203
21	0.0169	200	0.00450	1000	0.00173
22	0.0167	300	0.00300	10000	0.00
25	0.0145	500	0.00179		
30	0.00923	800	0.00112	$\epsilon$	qe10
35	0.00625	1000	0.000900	(eV)	( $10^{-16}\text{cm}^2$ )
40	0.00445	10000	0.00		
50	0.00252			13.27	0.00
60	0.00160	$\epsilon$	qe8	14	0.00450
80	0.000780	(eV)	( $10^{-16}\text{cm}^2$ )	15	0.0107
100	0.000444			16	0.0162
150	0.000160	13.15	0.00	18	0.0260
180	0.000100	14	0.00200	19	0.0284
200	0.00	15	0.00560	20	0.0298
		16	0.00920	22	0.0310
$\epsilon$	qe6	18	0.0156	25	0.0295
(eV)	( $10^{-16}\text{cm}^2$ )	19	0.0174	30	0.0250
		20	0.0181	35	0.0207
13.08	0.00	21	0.0181	40	0.0178
14	0.00550	22	0.0176	50	0.0138
15	0.0123	25	0.0155	60	0.0110
16	0.0190	30	0.0120	80	0.00780
18	0.0275	35	0.00950	100	0.00600
19	0.0294	40	0.00780	150	0.00370
20	0.0300	50	0.00560	200	0.00264
21	0.0298	60	0.00425	300	0.00164
22	0.0292	80	0.00275	500	0.000890
25	0.0255	100	0.00198	800	0.000510
30	0.0173	150	0.00109	1000	0.000390
35	0.0119	200	0.000710	10000	0.00
40	0.00850	300	0.000390		
50	0.00500	500	0.000181	$\epsilon$	qe11
60	0.00318	740	0.000100	(eV)	( $10^{-16}\text{cm}^2$ )
80	0.00157	800	0.00		
100	0.000910			13.30	0.00
150	0.000340	$\epsilon$	qe9	14	0.00155
200	0.000167	(eV)	( $10^{-16}\text{cm}^2$ )	15	0.00390
248	0.000100			16	0.00620
300	0.00	13.17	0.00	18	0.0111
		14	0.00360	19	0.0134
$\epsilon$	qe7	15	0.00810	20	0.0151
(eV)	( $10^{-16}\text{cm}^2$ )	16	0.0124	21	0.0162
		18	0.0184	22	0.0170
13.09	0.00	19	0.0194	25	0.0182
14	0.00230	20	0.0198	30	0.0185
15	0.00520	22	0.0196	35	0.0176
16	0.00820	25	0.0190	40	0.0162
18	0.0138	30	0.0179	50	0.0133
20	0.0184	35	0.0167	60	0.0110

Argon 3

Ar

80	0.00820	500	0.00355	13.98	0.00
100	0.00657	800	0.00219	14	0.000800
150	0.00435	1000	0.00172	15	0.0140
200	0.00322	10000	0.00	16	0.0275
300	0.00212			18	0.0470
500	0.00126	$\epsilon$	qe14	19	0.0500
800	0.000780	(eV)	( $10^{-16}\text{cm}^2$ )	20	0.0508
1000	0.000630			22	0.0490
10000	0.00	13.84	0.00	25	0.0420
		14	0.00300	30	0.0280
$\epsilon$	qe12	15	0.0205	35	0.0185
(eV)	( $10^{-16}\text{cm}^2$ )	16	0.0380	40	0.0129
		18	0.0600	50	0.00700
13.33	0.00	19	0.0650	60	0.00430
14	0.00200	20	0.0660	80	0.00197
15	0.00460	22	0.0640	100	0.00110
16	0.00600	25	0.0560	150	0.000370
18	0.00740	30	0.0390	200	0.000170
19	0.00770	35	0.0265	242	0.000100
20	0.00783	40	0.0193	300	0.00
22	0.00780	50	0.0115		
25	0.00730	60	0.00750	$\epsilon$	qe17
30	0.00611	80	0.00385	(eV)	( $10^{-16}\text{cm}^2$ )
35	0.00485	100	0.00225		
40	0.00395	150	0.000890	14.01	0.00
50	0.00283	200	0.000450	15	0.00830
60	0.00216	300	0.000175	16	0.0143
80	0.00140	380	0.000100	18	0.0203
100	0.00100	500	0.00	19	0.0213
150	0.000550			20	0.0216
200	0.000360	$\epsilon$	qe15	22	0.0214
300	0.000195	(eV)	( $10^{-16}\text{cm}^2$ )	25	0.0203
470	0.000100			30	0.0168
500	0.00	13.90	0.00	35	0.0130
		14	0.00150	40	0.0105
$\epsilon$	qe13	15	0.0160	50	0.00720
(eV)	( $10^{-16}\text{cm}^2$ )	16	0.0300	60	0.00540
		18	0.0475	80	0.00334
13.48	0.00	19	0.0510	100	0.00230
14	0.00250	20	0.0524	150	0.00118
15	0.00800	22	0.0520	200	0.000730
16	0.0135	25	0.0475	300	0.000375
18	0.0235	30	0.0360	500	0.000160
19	0.0275	35	0.0245	660	0.000100
20	0.0307	40	0.0176	800	0.00
21	0.0332	50	0.0100		
22	0.0352	60	0.00640	$\epsilon$	qe18
25	0.0387	80	0.00310	(eV)	( $10^{-16}\text{cm}^2$ )
30	0.0411	100	0.00177		
35	0.0410	150	0.000650	14.06	0.00
40	0.0400	200	0.000315	15	0.0105
50	0.0362	300	0.000113	16	0.0220
60	0.0317	315	0.000100	17	0.0283
80	0.0240	500	0.00	18	0.0315
100	0.0190			19	0.0328
150	0.0124	$\epsilon$	qe16	20	0.0332
200	0.00920	(eV)	( $10^{-16}\text{cm}^2$ )	21	0.0326
300	0.00605			22	0.0315

Argon 4

Ar

25	0.0260	200	0.0316	$\epsilon$	qe23
30	0.0193	300	0.0250	(eV)	( $10^{-16}\text{cm}^2$ )
35	0.0152	500	0.0180	14.30	0.00
40	0.0123	800	0.0130	15	0.00320
50	0.00860	1000	0.0110	16	0.00730
60	0.00640	10000	0.00	18	0.0145
80	0.00405			20	0.0199
100	0.00280	$\epsilon$	qe21	22	0.0240
150	0.00146	(eV)	( $10^{-16}\text{cm}^2$ )	25	0.0290
200	0.000910	14.21	0.00	30	0.0355
300	0.000475	15	0.00450	35	0.0405
500	0.000205	16	0.0110	40	0.0440
780	0.000100	18	0.0210	50	0.0480
800	0.00	19	0.0235	60	0.0495
		20	0.0248	80	0.0495
$\epsilon$	qe19	21	0.0250	100	0.0477
(eV)	( $10^{-16}\text{cm}^2$ )	22	0.0247	150	0.0395
14.09	0.00	25	0.0225	200	0.0330
15	0.00220	30	0.0170	300	0.0260
16	0.00500	35	0.0122	500	0.0192
18	0.00950	40	0.00940	800	0.0147
20	0.0123	50	0.00595	1000	0.0129
22	0.0142	60	0.00415	10000	0.00
25	0.0164	80	0.00233		
30	0.0191	100	0.00148	$\epsilon$	qe24
35	0.0208	150	0.000660	(eV)	( $10^{-16}\text{cm}^2$ )
40	0.0222	200	0.000370	14.71	0.00
50	0.0239	300	0.000163	15	0.00200
60	0.0245	385	0.000100	16	0.0100
80	0.0243	500	0.00	18	0.0300
100	0.0230			20	0.0480
150	0.0200	$\epsilon$	qe22	22	0.0650
200	0.0175	(eV)	( $10^{-16}\text{cm}^2$ )	25	0.0910
300	0.0140	14.23	0.00	30	0.124
500	0.0102	15	0.00400	35	0.150
800	0.00750	16	0.00960	40	0.165
1000	0.00640	18	0.0190	50	0.178
10000	0.00	19	0.0228	60	0.178
		20	0.0255	80	0.165
$\epsilon$	qe20	22	0.0285	100	0.146
(eV)	( $10^{-16}\text{cm}^2$ )	25	0.0310	150	0.112
14.15	0.00	30	0.0310	200	0.0930
15	0.00400	35	0.0290	300	0.0690
16	0.00900	40	0.0265	500	0.0470
18	0.0175	50	0.0215	800	0.0330
20	0.0230	60	0.0180	1000	0.0280
22	0.0270	80	0.0138	10000	0.00
25	0.0315	100	0.0110		
30	0.0362	150	0.00760	$\epsilon$	qe25
35	0.0400	200	0.00580	(eV)	( $10^{-16}\text{cm}^2$ )
40	0.0425	300	0.00395	15.20	0.00
50	0.0455	500	0.00242	16	0.00900
60	0.0460	800	0.00157	18	0.0320
80	0.0450	1000	0.00127	20	0.0630
100	0.0430	10000	0.00	22	0.0850
150	0.0367				

Argon 5

Ar

25	0.112	350	1.81
30	0.148	400	1.68
35	0.170	500	1.46
40	0.183	600	1.30
50	0.193	700	1.16
60	0.193	800	1.06
80	0.173	1000	0.916
100	0.152	10000	0.00
150	0.115		
200	0.0950		
300	0.0710		
500	0.0480		
800	0.0330		
1000	0.0270		
10000	0.00		

$\epsilon$ (eV)	$q_i$ ( $10^{-16}\text{cm}^2$ )
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15.76	0.00
16	0.0202
17	0.134
18	0.294
19	0.460
20	0.627
21	0.787
22	0.933
23	1.06
24	1.18
25	1.30
26	1.41
28	1.60
30	1.80
32	1.96
34	2.11
36	2.24
38	2.33
40	2.39
42.5	2.45
45	2.49
50	2.53
55	2.60
60	2.66
65	2.73
70	2.77
75	2.82
80	2.84
85	2.85
90	2.86
100	2.85
120	2.81
130	2.76
150	2.68
160	2.62
180	2.52
200	2.39
250	2.17
300	1.98

Argon 6



Numbers of References  
on Electron and Photon Collisions  
with Atoms and Molecules  
published in the 20th Century

Atoms (17)			Molecules (51)					
A + e,	A + hν			M + e,	M + hν,			
He	2	2170 *	2	H <sub>2</sub> , D <sub>2</sub>	1850	5	CH <sub>4</sub>	750
Ne	10	1140 *		N <sub>2</sub>	2140		CF <sub>4</sub>	390
Ar	18	1960		O <sub>2</sub>	1690		CCl <sub>4</sub>	210
Kr	36	790		CO	1190		CCl <sub>2</sub> F <sub>2</sub>	250
Xe	54	940		NO	850		CH <sub>3</sub> Cl	90
							SiH <sub>4</sub>	230
Li	3	450		F <sub>2</sub>	170		SiF <sub>4</sub>	140
Na	11	800		Cl <sub>2</sub>	320		GeH <sub>4</sub>	50
K	19	370		Br <sub>2</sub>	130			
Rb	37	220		I <sub>2</sub>	240			
Cs	55	370				6	C <sub>2</sub> H <sub>4</sub>	370
				HF	260		CH <sub>3</sub> OH	240
O	8	390		HCl	320			
				HBr	190			
F	9	90		HI	100	7	SF <sub>6</sub>	780
Cl	17	130						
			3	CO <sub>2</sub>	1160	8	C <sub>2</sub> H <sub>6</sub>	260
				H <sub>2</sub> O	850		C <sub>2</sub> F <sub>6</sub>	150
				O <sub>3</sub>	480		Si <sub>2</sub> H <sub>6</sub>	70
				N <sub>2</sub> O	450			
				NO <sub>2</sub>	300	9	C <sub>3</sub> H <sub>6</sub>	120
				H <sub>2</sub> S	270		C <sub>2</sub> H <sub>5</sub> OH	60
				SO <sub>2</sub>	260			
				CS <sub>2</sub>	240			
				OCS	240	11	C <sub>3</sub> H <sub>8</sub>	190
							C <sub>3</sub> F <sub>8</sub>	100
			4	C <sub>2</sub> H <sub>2</sub>	390	12	C <sub>4</sub> F <sub>8</sub>	100
				NH <sub>3</sub>	400		C <sub>6</sub> H <sub>6</sub>	240
				NF <sub>3</sub>	110		C <sub>6</sub> F <sub>6</sub>	100
				BF <sub>3</sub>	110			
				BCl <sub>3</sub>	90	60	C <sub>6</sub> O	300
				PH <sub>3</sub>	80			
				H <sub>2</sub> CO	180		r + v	700

\* He(Ne) + e only. Not include He(Ne) + hν papers.

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