

§4. Synthesis of Er₂O₃/buffer Multilayered Ceramic Coating for an Advanced Breeding Blanket Systems

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Erbium oxide (Er₂O₃) was shown to be the promising one of the candidate oxide coating materials to prevent magneto-hydrodynamic (MHD) pressure drop because of its high stability in liquid lithium and high electrical resistivity from the results of Er₂O₃ bulk and Physical Vapor Deposition (PVD) thin film. Furthermore, Er₂O₃ is also known to be a candidate for the tritium barrier coating. We have been applied Metal Organic Chemical Vapor Deposition (MOCVD) process to form large area oxide coating into duct tubing. We succeeded to make Er₂O₃ coating layer on metal substrate using MOCVD process. We found that the hydrogen permeation quantity was decreased compared with that of stainless steel (SUS 316) substrate without coating and it suggested that Er₂O₃ coating layer was also effectively able to become one of the candidate materials as the hydrogen permeation barrier.

In order to enhance these properties of Er₂O₃ as the electrical insulator for MHD and hydrogen permeation barrier coating, it is necessary to form thicker and higher crystallinity of Er₂O₃ coating layer. However, the epitaxial growth like the chemical vapor deposition is difficult to form thicker coating layer. We approached and investigated to form the double stacking coating layer on the SUS substrate using intermediate layer (buffer layer) for the thicker and high crystallinity of Er₂O₃ coating formation. Generally, it is well known that the crystallinity and alignment of coating layer affected by the lattice constant of the substrate. It is possible to the high crystallinity of Er₂O₃ coating by the buffer layer formation with similar Er₂O₃ lattice constant between Er₂O₃ and substrate. The relationship of lattice constant between coating and substrate is often discussed using mismatch factor. The mismatch factor is estimated by the following formula;

$$\text{Mismatch factor } f = |(b-a)| / a$$

a: lattice constant of substrate,
b: lattice constant of coating

Table.1 The comparisons of lattice mismatch factor between various multilayered coating case

Sample	Er ₂ O ₃ Lattice parameter (Å)	Buffer Lattice parameter (Å)	SUS (γ-Fe) Lattice parameter (Å)	Lattice mismatch (f)
Non-Buffer				1.86
CeO ₂	10.55	5.41	3.646	0.95/0.48
Y ₂ O ₃		10.60		0.005/1.91

The consistence between coating layer and metal substrate intend to be better with similar to the mismatch factor. In this study, Y₂O₃ and CeO₂ having the similar lattice constant of Er₂O₃ and SUS materials were selected as the buffer layer. Table.1 indicates the summary of mismatch factor in this study.

Typical image and X-ray diffraction of the Er₂O₃/CeO₂ multilayered coating sample are shown in Fig.1. We found that the interference band was decreased compared with no buffer coating sample. It suggested that coating thickness uniformity was improved by the buffer layer on substrate. From the X-ray diffraction, main diffraction peaks of Er₂O₃ and CeO₂ phases were also observed sharply. We concluded that Er₂O₃ coating layer was easily formed on CeO₂ buffer layer by MOCVD, and buffer layer was effective to form homogeneous coating layer. TEM image of the cross-sectional area in on the Er₂O₃/CeO₂ multilayered ceramic coating on the stainless steel substrate is shown in Fig.2. Er₂O₃ crystal formed columnar-like structure and it formed densely compared with the sample without CeO₂ buffer layer. The origin of grain growth of Er₂O₃ crystal seemed to be CeO₂ crystals. In the future, the microstructure of Er₂O₃/Y₂O₃ multilayered coating samples will be also investigated.

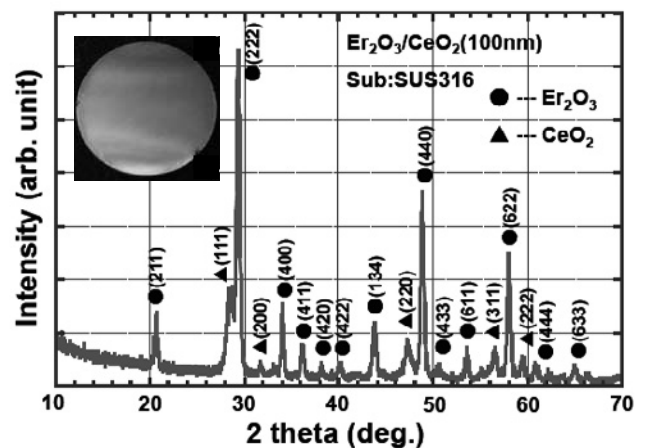


Fig.1 Typical X-Ray diffraction of the Er₂O₃/CeO₂ multilayered ceramic coating on the stainless steel (SUS 316) substrate.

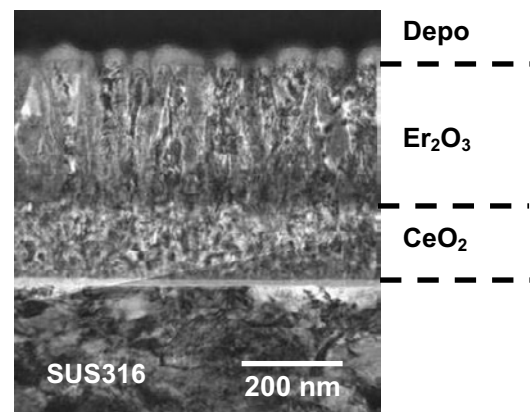


Fig.2 TEM image of cross-section area on the Er₂O₃/CeO₂ multilayered ceramic coating on the stainless steel (SUS 316) substrate.