

§3. Depth Profile of Deuterium in Fe_2O_3 under Low-Energy Deuterium Exposure

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Oxide layers such as Fe_2O_3 will be formed on stainless steel (SUS), one of structure materials widely used for fusion devices [1], during removal process of hydrogen isotopes [2]. Thus, the depth distribution of deuterium (D) and amount of D-retention in the oxide layers are important for designing the devices. We have studied the depth profile of D and D-retention in Fe_2O_3 under low-energy D plasma exposure [3].

Fe_2O_3 layers were prepared by deposition of Fe on SiO_2 -glass substrates followed by oxidation in air at 400 to 500 °C for more than 5 min. X-ray diffraction (XRD) shows that the oxide is $\alpha\text{-Fe}_2\text{O}_3$, i.e., hematite (hexagonal corundum structure) and the composition is evaluated to be Fe:O=2:3 with an accuracy of 10% by Rutherford backscattering spectroscopy (RBS) of 1.8 MeV He^+ . RBS is also employed to evaluate the film thickness. Nuclear reaction, $\text{D}(^3\text{He},\alpha)\text{P}$, analysis (NRA) with 1 MeV $^3\text{He}^+$ was employed to obtain the depth profile of D. Here, the incident angle and detection angle of α -particles are 20° and 70° from the sample surface normal, respectively (NRA angle is 90° measured from the incident beam direction).

Figure 1 shows the depth profile of D in $\alpha\text{-Fe}_2\text{O}_3$. One sees that D's are distributed beyond the depth resolution (43 nm) and the range (30 nm) calculated [4] by considering the fractions of D_3 (60%), D_2 (38%) and D(2%) [5]. It also appears that more than 90 % of D's are located within the film thickness of 0.1 μm , indicating small diffusivity of D in $\alpha\text{-Fe}_2\text{O}_3$ near room temperature. By integrating the D density up to 0.2 μm , the amount of D-retention is obtained to be $45 \times 10^{15} \text{ cm}^{-2}$ and corrected to be $50 \times 10^{15} \text{ cm}^{-2}$ by taking $^3\text{He}^+$ beam induced desorption (evaluated from measurement of D-retention vs $^3\text{He}^+$ beam fluence) into account, in reasonable agreement with the reported value [3]. After the D-exposure, the XRD intensity decreased to 1/3 of that before D-exposure and the lattice parameter increased by 0.4 %. These modifications are not likely due to irradiation effects by low energy D (limited within 30 nm) but incorporation of D into Fe_2O_3 lattice.

In the mixed layers (more than 2 μm) of $\gamma\text{-Fe}_2\text{O}_3$ (maghemite, cubic spinel structure) and $\alpha\text{-Fe}_2\text{O}_3$, which were prepared by oxidation of Fe at 400 to 500 °C, the

depth profile of D appears to be similar to that in $\alpha\text{-Fe}_2\text{O}_3$, but D-density is larger than that in $\alpha\text{-Fe}_2\text{O}_3$ in regions deeper than 0.1 μm . D-retention is evaluated to be $73 \times 10^{15} \text{ cm}^{-2}$, by integrating the D-density up to 0.5 μm and correcting $^3\text{He}^+$ beam induced desorption, the value being in agreement with the reported value ($76 \times 10^{15} \text{ cm}^{-2}$ [3]). D-retention in the mixed layers is larger than that in $\alpha\text{-Fe}_2\text{O}_3$. The difference could be largely due to the difference of the D-distribution mentioned above. D-retention in $\alpha\text{-Fe}_2\text{O}_3$ is larger than that in Fe ($12 \times 10^{15} \text{ cm}^{-2}$) and SUS316L ($36 \times 10^{15} \text{ cm}^{-2}$). Measurements of D distribution in SUS and Fe, dynamic D-retention and thermal desorption of D are under way.

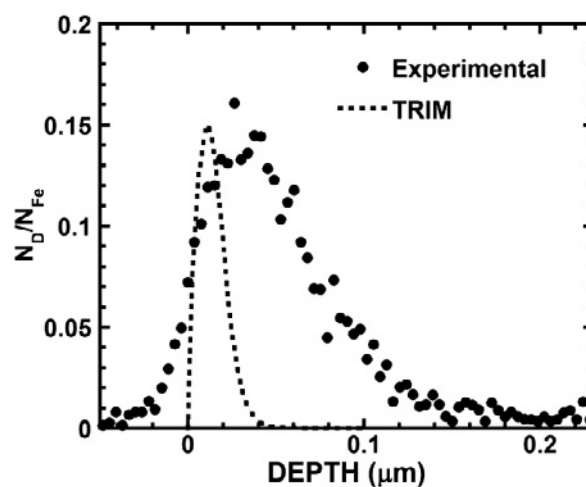


Fig. 1 Depth distribution of D density in $\alpha\text{-Fe}_2\text{O}_3$ (thickness of 100 nm) normalized to Fe density exposed to D-plasma (1.5 kV AC glow-discharge in 0.5 Torr D_2 for 30 min) (●). The dotted line indicates the range profile of D calculated using TRIM code [4].

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